

## Methyl 4-nitrophenyl sulfoxide: $\pi$ -stacked chains of hydrogen-bonded $R_2^2(10)$ dimers

Christopher Glidewell,<sup>a\*</sup> John N. Low,<sup>b</sup> Janet M. S. Skakle<sup>b</sup> and James L. Wardell<sup>c</sup>

<sup>a</sup>School of Chemistry, University of St Andrews, Fife KY16 9ST, Scotland,

<sup>b</sup>Department of Chemistry, University of Aberdeen, Meston Walk, Old Aberdeen AB24 3UE, Scotland, and <sup>c</sup>Departamento de Química Inorgânica, Instituto de Química, Universidade Federal do Rio de Janeiro, 21945-970 Rio de Janeiro, RJ, Brazil

Correspondence e-mail: cg@st-andrews.ac.uk

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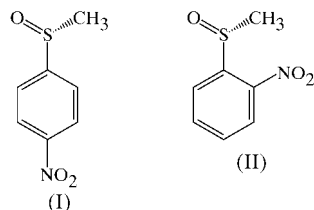
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Molecules of the title compound,  $C_7H_7NO_3S$ , are linked into centrosymmetric  $R_2^2(10)$  dimers by paired C—H $\cdots$ O hydrogen bonds, and these dimers are linked into [110] chains by a single aromatic  $\pi$ – $\pi$  stacking interaction.

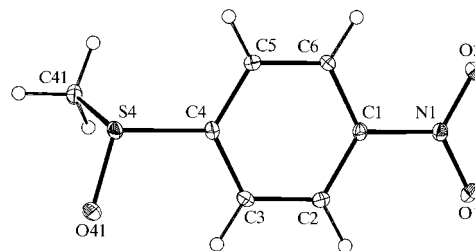
### Comment

Molecules of the title compound, (I), are chiral, but the centrosymmetric space group accommodates equal numbers of the *R* and *S* enantiomers; the selected reference molecule (Fig. 1) is of the *R* configuration. Within the molecule, the ring C—C distances span the very narrow range 1.384 (2)–1.392 (2) Å. The two independent C—S distances (Table 1) are very similar but with the distance to the aryl C marginally the longer of the two. The reference mean values for  $C_{\text{aryl}}\text{—S(O)}\text{—}$  and  $C_{\text{sp}^3}\text{—S(O)}\text{—}$  distances (Allen *et al.*, 1987) are 1.790 and 1.809 Å, respectively, while the reference mean value for the S=O distance in sulfoxides is 1.497 Å, so the S—O distance in (I) is entirely characteristic.

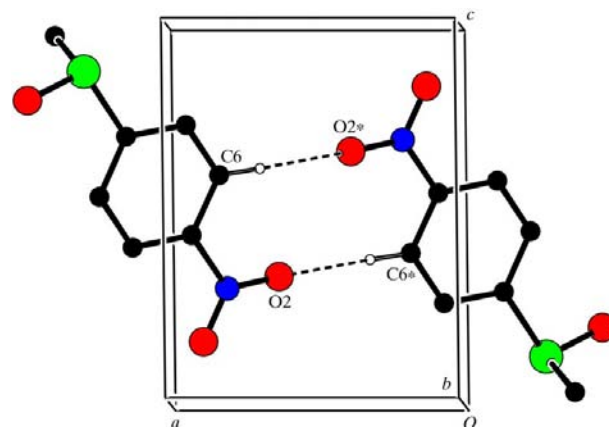


The internal C—C—C angles at C1 and C4 in (I), *ipso* to the electron-withdrawing  $\text{NO}_2$  and  $\text{S(O)CH}_3$  substituents, respectively, are both greater than  $120^\circ$ , as expected (Domenicano & Murray-Rust, 1979). At the pyramidal S atom, the bond angles involving sulfoxide atom O41 are both signifi-

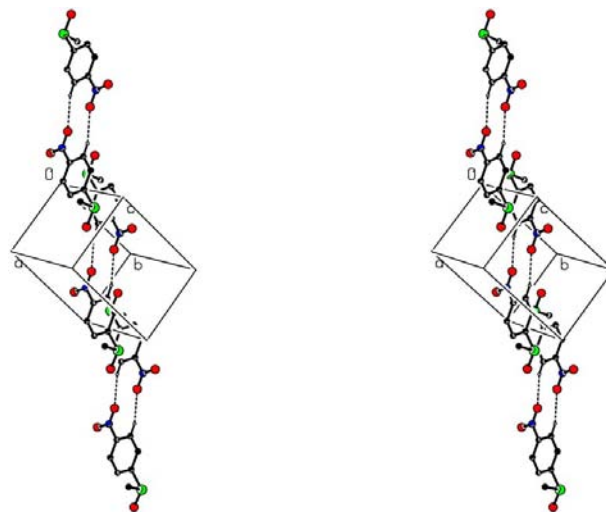
cantly larger than the C—S—C angle. The conformation of the  $\text{S(O)}\text{—CH}_3$  fragment, with the sulfonyl O atom almost coplanar with the aryl ring, may be controlled by an intra-



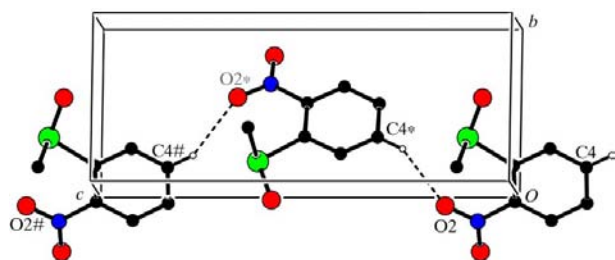
**Figure 1**  
The *R* enantiomer of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level.



**Figure 2**  
Part of the crystal structure of (I), showing the formation of a centrosymmetric  $R_2^2(10)$  dimer. For the sake of clarity, H atoms not involved in the motif shown have been omitted. Atoms marked with an asterisk (\*) are at the symmetry position  $(1-x, 1-y, 1-z)$ .



**Figure 3**  
Stereoview of part of the crystal structure of (I), showing the formation of a [110] chain of  $\pi$ -stacked  $R_2^2(10)$  dimers. For the sake of clarity, H atoms not involved in the motif shown have been omitted.



**Figure 4**

Part of the crystal structure of (II) (Ianelli *et al.*, 1992), showing the formation of a zigzag  $C(7)$  chain along  $[001]$ . Atomic coordinates and atom labels are as given in the original report. For the sake of clarity, H atoms not involved in the motif shown have been omitted. Atoms marked with an asterisk (\*) or a hash (#) are at the symmetry positions  $(x, \frac{1}{2} - y, \frac{1}{2} + z)$  and  $(x, y, 1 + z)$ , respectively.

molecular dipolar attraction involving the positively charged atom H3 and the negatively charged atom O41. The nitro group is essentially coplanar with the aryl group, and a similar conformation was found in the isomeric compound methyl 2-nitrophenyl sulfoxide, (II) (Ianelli *et al.*, 1992).

The molecules of (I) are weakly linked into centrosymmetric dimers by a nearly linear  $C-H \cdots O$  hydrogen bond, and these weak dimers are further linked into chains by a single aromatic  $\pi$ - $\pi$  stacking interaction. Ring atom C6 in the molecule at  $(x, y, z)$ , which is adjacent to the nitro group and hence has the most polarized  $C-H$  bond in the molecule, acts as hydrogen-bond donor to nitro atom O2 in the molecule at  $(1 - x, 1 - y, 1 - z)$ , so forming an  $R_2^2(10)$  dimer (Bernstein *et al.*, 1995) centred at  $(\frac{1}{2}, \frac{1}{2}, \frac{1}{2})$  (Table 2, Fig. 2). The aryl rings of the molecules at  $(x, y, z)$  and  $(2 - x, 2 - y, 1 - z)$ , which are components of the  $R_2^2(10)$  dimers centred, respectively, at  $(\frac{1}{2}, \frac{1}{2}, \frac{1}{2})$  and  $(\frac{3}{2}, \frac{3}{2}, \frac{1}{2})$ , are parallel, with an interplanar spacing of  $3.367(2) \text{ \AA}$ . The centroid-centroid separation is  $3.666(2) \text{ \AA}$ , corresponding to a nearly ideal centroid offset of  $1.450(2) \text{ \AA}$ . Propagation by inversion of this stacking interaction then generates a chain of  $\pi$ -stacked dimers running parallel to the  $[110]$  direction (Fig. 3).

The supramolecular structure of (II) was not discussed at all in the original report (Ianelli *et al.*, 1992). However, use of the atom coordinates for (II) retrieved from the Cambridge Structural Database (Allen, 2002; refcode KONKON) shows that, in contrast with the supramolecular structure of (I), that of (II) consists of isolated  $C(7)$  zigzag chains built from a single  $C-H \cdots O$  hydrogen bond and generated by a glide plane in space group  $P2_1/c$  (Fig. 4).  $\pi$ - $\pi$  stacking interactions are absent from the structure of (II).

## Experimental

A sample of compound (I) was prepared by oxidation of a commercial sample of methyl 4-nitrophenyl sulfide (Aldrich), using hydrogen peroxide and  $TiCl_3$ , following the published procedure of Watanabe *et al.* (1981). Crystals of (I) suitable for single-crystal X-ray diffraction were grown by slow evaporation of a solution in ethanol (m.p. 430–432 K).

## Crystal data

$C_7H_7NO_3S$   
 $M_r = 185.20$   
 Triclinic,  $P\bar{1}$   
 $a = 6.1522(2) \text{ \AA}$   
 $b = 7.8550(3) \text{ \AA}$   
 $c = 7.9939(3) \text{ \AA}$   
 $\alpha = 83.684(3)^\circ$   
 $\beta = 89.320(2)^\circ$   
 $\gamma = 88.430(3)^\circ$   
 $V = 383.81(2) \text{ \AA}^3$   
 $Z = 2$

$D_x = 1.603 \text{ Mg m}^{-3}$   
 Mo  $K\alpha$  radiation  
 Cell parameters from 1739 reflections  
 $\theta = 3.3\text{--}27.5^\circ$   
 $\mu = 0.38 \text{ mm}^{-1}$   
 $T = 120(2) \text{ K}$   
 Block, colourless  
 $0.35 \times 0.35 \times 0.30 \text{ mm}$

## Data collection

Nonius KappaCCD area-detector diffractometer  
 $\varphi$  scans, and  $\omega$  scans with  $\kappa$  offsets  
 Absorption correction: multi-scan (SORTAV; Blessing, 1995, 1997)  
 $T_{\min} = 0.878$ ,  $T_{\max} = 0.894$   
 6313 measured reflections

1739 independent reflections  
 1588 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.038$   
 $\theta_{\max} = 27.5^\circ$   
 $h = -7 \rightarrow 8$   
 $k = -10 \rightarrow 9$   
 $l = -10 \rightarrow 10$

## Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.030$   
 $wR(F^2) = 0.079$   
 $S = 1.11$   
 1739 reflections  
 110 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.033P)^2 + 0.1817P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$   
 $\Delta\rho_{\max} = 0.32 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\min} = -0.41 \text{ e \AA}^{-3}$

**Table 1**

Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ).

C4—S4	1.8003 (14)	S4—C41	1.7909 (15)
S4—O41	1.4937 (11)		
C2—C1—C6	123.20 (13)	C4—S4—O41	106.29 (6)
C3—C4—C5	121.67 (13)	O41—S4—C41	107.10 (7)
C4—S4—C41	96.61 (6)		
C3—C4—S4—O41	6.56 (13)	C2—C1—N1—O1	−0.87 (19)
C3—C4—S4—C41	−103.45 (12)	C2—C1—N1—O2	178.88 (12)

**Table 2**

Geometry of hydrogen bonds and short intramolecular contacts ( $\text{\AA}$ ,  $^\circ$ ).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
C3—H3 $\cdots$ O41	0.95	2.49	2.907 (2)	106
C6—H6 $\cdots$ O2 <sup>i</sup>	0.95	2.56	3.478 (2)	164

Symmetry code: (i)  $1 - x, 1 - y, 1 - z$ .

Crystals of (I) are triclinic. Space group  $P\bar{1}$  was selected and confirmed by the successful structure analysis. All H atoms were located from difference maps and subsequently treated as riding atoms, with  $C-H$  distances of 0.95 (aromatic) and 0.98  $\text{\AA}$  (methyl).

Data collection: *KappaCCD Server Software* (Nonius, 1997); cell refinement: *DENZO-SMN* (Otwinowski & Minor, 1997); data reduction: *DENZO-SMN*; program(s) used to solve structure: *OSCAIL* (McArdle, 2003) and *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *OSCAIL* and *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97* and *PRPKAPPA* (Ferguson, 1999).

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: SK1686). Services for accessing these data are described at the back of the journal.

## References

- Allen, F. H. (2002). *Acta Cryst.* **B58**, 380–388.
- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). *Angew. Chem. Int. Ed. Engl.* **34**, 1555–1573.
- Blessing, R. H. (1995). *Acta Cryst.* **A51**, 33–38.
- Blessing, R. H. (1997). *J. Appl. Cryst.* **30**, 421–426.
- Domenicano, A. & Murray-Rust, P. (1979). *Tetrahedron Lett.* **20**, 2283–2286.
- Ferguson, G. (1999). *PRPKAPPA*. University of Guelph, Canada.
- Ianelli, S., Musatti, A., Nardelli, M., Benassi, R., Folli, U. & Taddei, F. (1992). *J. Chem. Soc. Perkin Trans. 2*, pp. 49–57.
- McArdle, P. (2003). *OSCAIL for Windows*. Version 10. Crystallography Centre, Chemistry Department, NUI Galway, Ireland.
- Nonius (1997). *KappaCCD Server Software*. Windows 3.11 Version. Nonius BV, Delft, The Netherlands.
- Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography*, Part A, edited by C. W. Carter Jr & R. M. Sweet, pp. 307–326. New York: Academic Press.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
- Watanabe, Y., Numata, T. & Oae, S. (1981). *Synthesis*, pp. 204–206.